1 Marks Questions
1. Soaps are formed by the saponification of
(a) Alcohols
(b) simple ester
(c) carboxylic acids
(d) glycerides
Ans. (d) glycerides
2. The functional group of butanone is
(a) Carboxyl
(b) ketonic
(c) aldehydic
(d) alcoholic
Ans. (b) ketonic
3. Enzyme which converts starch into glucose is
(a) Zymase
(b) Maltase
(c) Diastase
(d) Invertase
Ans. (a) Zymase
4. The first compound to be prepared in the laboratory was
(a) Methane
(b) Ethyl alcohol
(c) acetic acid 20:48
(d) Urea.
Ans. (d) Urea.
5. The IUPAC name of CH ₃ CHO is
(a) Acetaldehyde
(b) Formaldehyde
(c) Methyl formaldehyde
(d) Ethanal.
Ans. (d) Ethanal.
6. Rectified spirit is
(a) 50% ethanol
(b) 80% ethanol
(c) 95% ethanol

(d) 40 to 50% ethanol Ans. (c) 95% ethanol

(a) an oxidising agent (b) a reducing agent (c) a bleaching agent (d) none of these

Ans. (a) an oxidising agent

(a) Isoprene

7. Dilute alkaline \mathbf{KMnO}_4 solution is

8. The by product in soap industry is

(a) Isoprene (b) Ethylene glycol (c) glycerol (d) butane Ans. (c) glycerol 20:48 🗸 9. An example of soap is (a) C₁₅H₃₁COONa (b) CH, COONa (c) C.H.COONa (d) C1.H1.OSO1.Na Ans. (a) $C_{15}H_{31}COONa$ 10. The number of C-H bonds in ethane C, H, molecule are (a) 4 (b) 6 (c) 8 (d) 10 Ans. (b) 6 11. The odour of acetic acid resembles that of (b) Burning Plastic (c) Vinegar (d) Kerosene Ans. (c) Vinegar 12. Diamond is not a good conductor of electricity because (a) It is very hard (b) Its structure is very compact (c) It is not soluble in water (d) It has no free electrons to conduct electric current. Ans. (d) It has no free electrons to conduct electric current. 13. Alcohols can be produced by the hydration of (a) Alkenes (b) alkynes (c) alkanes 20:48 🗸

Ans. (a) Alkenes

(d) acids

14. The IUPAC name of CH₃CHO is

8. The by product in soap industry is

- (a) Acetaldehyde
- (b) formaldehyde
- (c) methyl formaldehyde
- (d) ethanol
- Ans. (d) ethanol

(c) alkanes	
(d) acids 20.48	Q
Ans. (a) Alkenes	
14. The IUPAC name of CH ₃ CHO is	
(a) Acetaldehyde	
(b) formaldehyde	
(c) methyl formaldehyde	
(d) ethanol	
Ans. (d) ethanol	
15. IUPAC name of first member of homologous series of ketones is	
(a) Ethanone	
(b) methanone	
(c) Propanone	
(d) Butanone	
Ans. (c) Propanone	
16. An unknown compound has the smell of vinegar. Identify it.	
Ans. Acetic acid	
17. Out of butter and groundnut oil which is unsaturated in nature?	
Ans. Groundnut oil	
18. Which has triple bond, C_2H_4, C_3H_4, C_3H_6	
Ans. C_3H_4	
19. Which substance is added for the denaturation of ethyl alcohol	
Ans. Methyl Alcohol	
20. Which ions are responsible for making water hard?	o fi
	SE 22
Ans. Ca^{1-} and Mg^{2-}	
24 Pahana adah aka malambar farmusia C TI hara	
21. Ethane, with the molecular formula C ₂ H ₆ has	
(a) 6 covalent bonds	
(b) 7 covalent bonds	
(c) 8 covalent bonds	
(d) 9 covalent bonds	
Ana. (b) 7 covalent bonds	

22. Butanone is a four carbon compound with the functional group

(a) carboxylic acid
(b) aldehyde
(c) ketone
(d) alcohol
Ans. (c) Ketone

13. Alcohols can be produced by the hydration of

(a) Alkenes(b) alkynes

15. IUPAC name of first member of homologous series of ketones is (a) Ethanone (b) methanone (c) Propanone (d) Butanone Ans. (c) Propanone 16. An unknown compound has the smell of vinegar. Identify it. Ans. Acetic acid 17. Out of butter and groundnut oil which is unsaturated in nature? Ans. Groundnut oil 18. Which has triple bond, C_2H_4 , C_3H_4 , C_3H_6 Ans. C.H. 19. Which substance is added for the denaturation of ethyl alcohol Ans. Methyl Alcohol 20. Which ions are responsible for making water hard? Ans. Ca2- and Mg2-21. Ethane, with the molecular formula C, H, has (a) 6 covalent bonds (b) 7 covalent bonds (c) 8 covalent bonds (d) 9 covalent bonds Ana. (b) 7 covalent bonds 22. Butanone is a four carbon compound with the functional group (a) carboxylic acid (b) aldehyde (c) ketone (d) alcohol Ans. (c) Ketone 23. While cooking, if the bottom of the vessels is getting blackened on the outside, it means that (a) the fuel is not cooked completely. (b) the fuel is not burning completely. (c) the fuel is wet. (d) the is burning completely. Ans. (b) the fuel is not burning completely. 24. Which of the following hydrocarbons undergo addition reactions?

C2H6, C3H6, C3H6, C2H2and CH

Ans. C. H. and C.H. will undergo addition reactions.