Pragati Education.
Carbon and it's Compounds
1. Give the names of the functional group;-
(i) -CHO (ii) -C=O (iii) -OH (iv) -COOH
Ans- (i) Aldehyde group (ii) Ketone group (iii) alcoholic (iv) carboxylic
2.Draw the electron dot structure of
a)H2O
<u>b)CO2</u>
<u>c)O2</u>
<u>d)N2</u>
<u>e)F2</u>
<u>F)CH4</u>
<u>G)C2H6</u>
<u>H)Cl2</u>
3.why carbon is versatile in nature?
4. Give 4 difference in diamond and graphite? Define covalency?
Ans-The number of electrons contributed by an atom in the formation of a covalent bond is called the covalency.
The covalency of an element in a covalent compound is equal to the number of covalent bond formed by an atom
of the element with other atom.
5. What is the formula of an alkyne
having five carbon atoms?
6. Write the name of the functional
group present in the following
compounds.
(i) CH3CH2OH
(ii) CH 3CH2CI
(iii) Ethanal

(iv) Acetic acid

(v) Hexyne

7.Name the 4 th member of ketone homologous series.